ME597/PHYS57000 Fall Semester 2009 Lecture 26

Scanning Probe Nanolithography

- STM early work
- Arranging atoms with a tip
- Local Oxidation Lithography (Electrochemical)
- Dip Pen Lithography
- Nanografting

Recent Topical Review:

Garcia, Martinez and Martinez, Chem. Soc. Rev. **35**, 29 (2006) Rosa and Liang, J. Phys.: Condens. Matter **21**, 483001 (2009)

STM-based Results

Writing nanometer-scale symbols in gold using the scanning tunneling microscope

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(Received 5 December 1988; accepted for publication 30 January 1989)





65 nm x 40 nm x 2 nm

D. Eigler moves and controls position of an individual Xe atom (1989)



35 Xenon atoms on Ni(110)

D.M. Eigler and E.K. Schweizer, Nature 344, 524 (1990).

An atomic pencil!



Crommie et al., Science **262**, 218 (1993)

Physical Mechanism



Local Oxidation Lithography

Early Work:

- J. Dagata et al. Appl. Phys. Lett. 56, 2001 (1990).
- H.C. Day and D.R. Allee, Appl. Phys. Lett. 62, 2691 (1993).
- E.S. Snow et al, Appl. Phys. Lett. 72, 3071 (1998).
- R. Held, et. al., Appl. Phys. Lett. 73, 262 (1998).

Scanning Under Ambient Conditions



Upon approach, a water neck forms. Due to surface tension, an additional force is required to pull the tip from the substrate. The additional force depends on the shape (r_1, r_2) of the water meniscus.

Water layer adsorbed on quartz at room temperature



Pashley and Kitchener, J. Colloid and Interface Sci., 71, 491 (1979).

Water layer adsorbed on SiO_2 at room temperature



Asay and Kim, J. Phys. Chem B 109, 16761 (2005).

SPM Oxidation Lithography - Schematic



c) Feedback OFF -20 V for ~2 sec



d) non-contact, voltage OFF; feedback ON



Local Chemistry at the Tip

 Si rapidly reacts with oxygen to form passivating layer of SiO₂

 Water vapor also oxides Si; (gaseous O₂ is not required)

3. Large electric field dissociates water

 4. OH⁻ diffuses through SiO₂ faster than O₂, allowing the oxide to grow



Si

Examples - Local Oxidation Lithography



S.W. Howell, Sandia National Labs



P. Ares, Nanotec Electronica

Minimum Feature Size



Estimating the height:

$$h \sim \left(\frac{t}{t_o}\right)^{\gamma}$$

$$\gamma \sim 0.1 - 0.3$$

$$t \sim 0.005 - 1 \text{ s}$$



Tello and Garcia, Appl. Phys. Lett. 79, 424 (2001).

Line widths





Tello and Garcia, J. Appl. Phys. 92, 4075 (2002)

Chemical Writing





Nanolithography with Molecules Lateral force images of Au substrate; RH 35%



Piner, Zhu, Xu, Hong, Mirkin, Science 283, 661 (1999)



Fig. 1. Nanoscale molecular letters written on an Au(111) surface with MHA molecules by DPN.

115 words in 10 minutes

60 nm As soon as I mention this, people tell me about miniaturization, and how For it has progressed today. They tell me about electric motors that are the size of the not I an your small Finger. And there is a device on the market, they tell me, by which you can write the Lord's Prayer on the head of a pin. But that's nothing! that's the most primitive, halting step in the direction I intend to discuss. It is a staggeringly small world that is below. In the year 2000, when they lock back at this age, they will wonder why it was not until the year 1950 that anybody began seriously to move in this direction. 400 nm Richard P. Feynman, 1968

Nanografting

Early Work:

S. Xu and G.Y. Liu, "Nanometer-scale fabrication by simultaneous nanoshaving and molecular selfassembly", *Langmuir* 13, 127–29 (1997).

Self-Assembled Monolayers

Self-Assembled Monolayers (SAMs) are ordered molecular assemblies formed by the adsorption of an active surfactant onto a solid surface.

Early Work: R.G. Nuzzo and D.L. Allara, J. Am. Chem. Soc. 105, 4481 (1983).

Molecules that can form a SAM (schematic):



The Self-Assembly Process



Nanografting



Liu, et al., Annu. Revs. Phys. Chem. 59, 367 (2008).

Nanografting Thiols on Au



 $1.5 \ \mu m \times 1.5 \ \mu m$

Two nanografted spirals, 620 nm in diameter, with average line spacing of 40 nm, average line-widths (FWHM) of 15 nm, and average heights of 0.15 nm. Mono-atomic steps in Au(111) are clearly present.



Garcia, Martinez and Martinez, Chem. Soc. Rev. 35, 29 (2006)