

Nanomaterials

Lecture 1: Film Deposition Methods

Film Deposition Methods

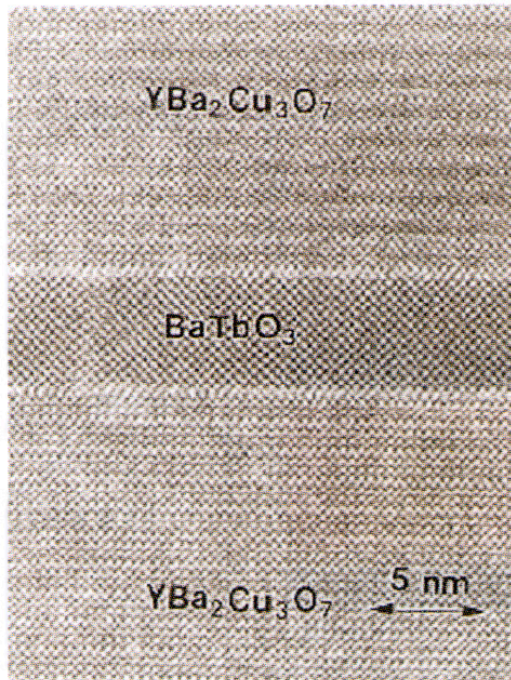


Figure 29: HRTEM micrograph of a sputter deposited trilayer system [19].

- Evaporation
- Molecular Beam Epitaxy
- Pulsed Laser Deposition
- Sputtering
- Chemical Vapor Deposition
- Atomic Layer Deposition

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Influence of Background Vacuum Pressure

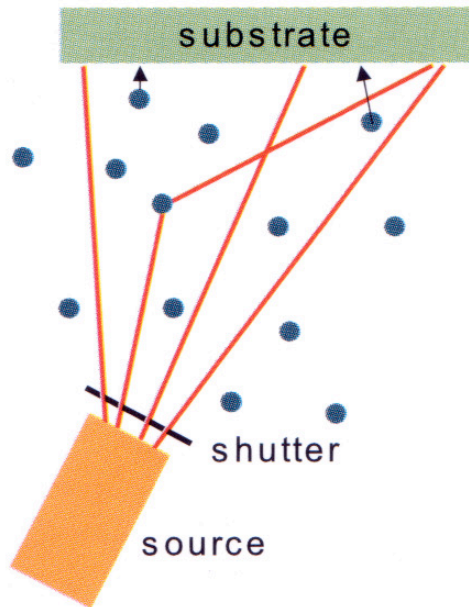


Figure 1: Schematics of a deposition system.

p (mbar)	Mean free path (cm)	Monolayers per sec
10^0	6.8×10^{-3}	3.3×10^5
10^{-3}	6.8×10^0	3.3×10^2
10^{-6}	6.8×10^3	3.3×10^{-1}
10^{-9}	6.8×10^6	3.3×10^{-4}

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Deposition into Trenches

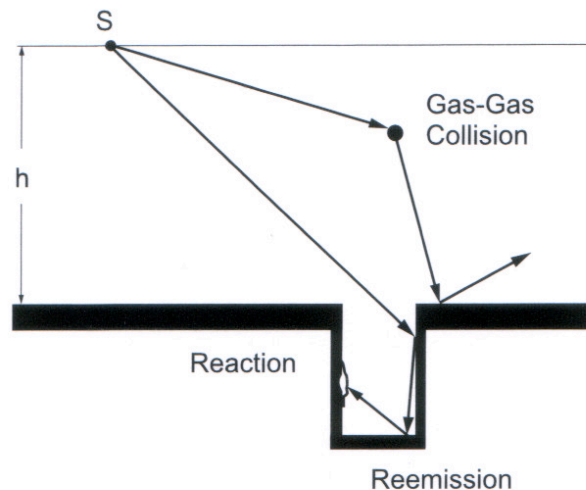


Figure 2: Deposition into a hole (after ref. 20). The example of an atom with starting position, S, with a height, h, above the substrate illustrates that either gas-gas collisions or reflection from the surface are necessary for a uniform deposition on bottom and sidewalls of a hole.

At low pressure, the mean free path can exceed the trench dimensions.

Consequently, surface reflections are necessary to achieve uniform deposition on the bottom and sidewalls of a trench.

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Elements of Surface Morphology

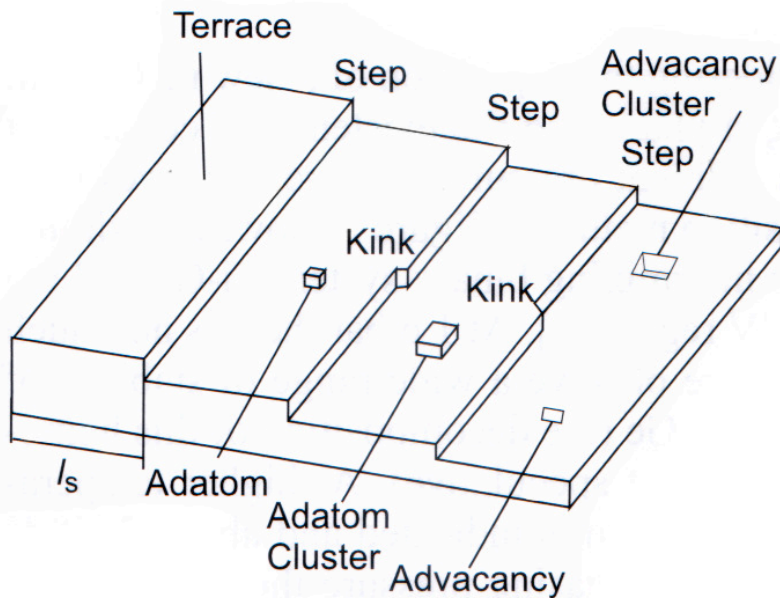


Figure 5: Schematic view of the elements of the surface morphology [3].

Nucleation and growth of a film proceeds from energetically favorable points on the substrate surface.

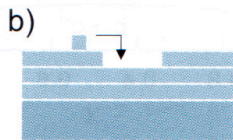
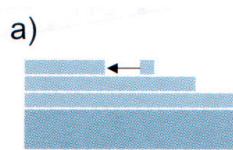
Even the cleanest polished surface shows some structure.

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Homoepitaxy

Epitaxy: Growth of a film with the same crystal orientation as the substrate → “Single crystal on a single crystal”

Homoepitaxy: Film and substrate are the same material



Growth modes of homoepitaxy:

- (a) Step-propagation
- (b) 2-D island growth
- (c) Multi-layer growth

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Heteroepitaxy

Heteroepitaxy: Film and substrate are different materials

Critical materials parameters that dictate growth mode:

- (1) Surface energy, γ
- (2) Lattice constants (lattice match)

If there is a lattice mismatch, there is a competition between strain energy ($\sim d^3$) and surface energy ($\sim d^2$):

$$\Delta W = W_{\text{surf}} + W_{\text{relax}} = C_1 \gamma d^2 - C_2 k \xi^2 d^3$$

where k = bulk modulus, ξ = strain

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Lattice Mismatched Growth Mode

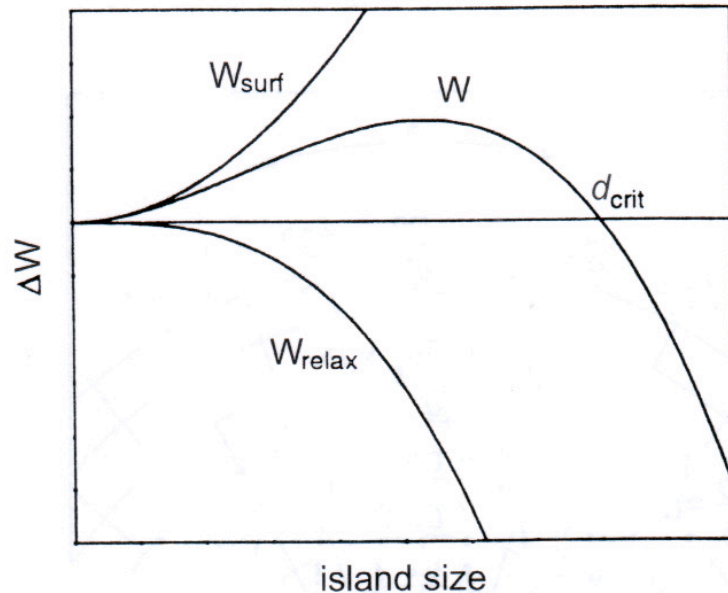


Figure 9: Energy contributions as a function of the island size [5].

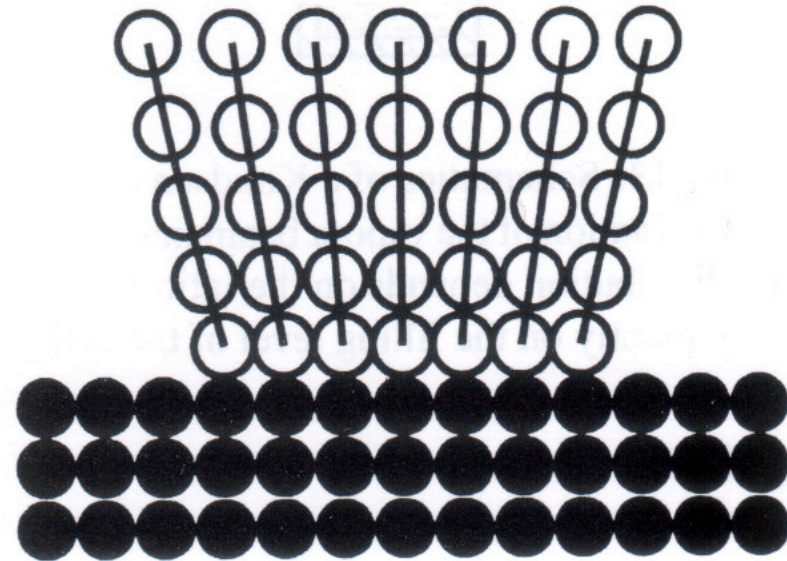
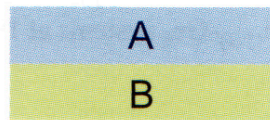


Figure 10: Strain relaxation in pseudo-morphic (dislocation - free) islands.

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Classes of Heteroepitaxy Growth Modes

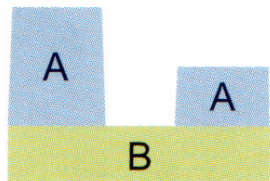
(a)



Frank-van der Merve

Lattice matched, $\gamma_{\text{layer}} + \gamma_{\text{substrate/layer}} < \gamma_{\text{substrate}}$
→ Perfect wetting and layer by layer growth

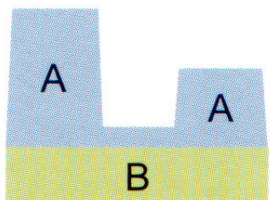
(b)



Volmer-Weber

Lattice matched, $\gamma_{\text{layer}} + \gamma_{\text{substrate/layer}} > \gamma_{\text{substrate}}$
→ Island growth

(c)



Stranski-Krastanov

Lattice mismatched

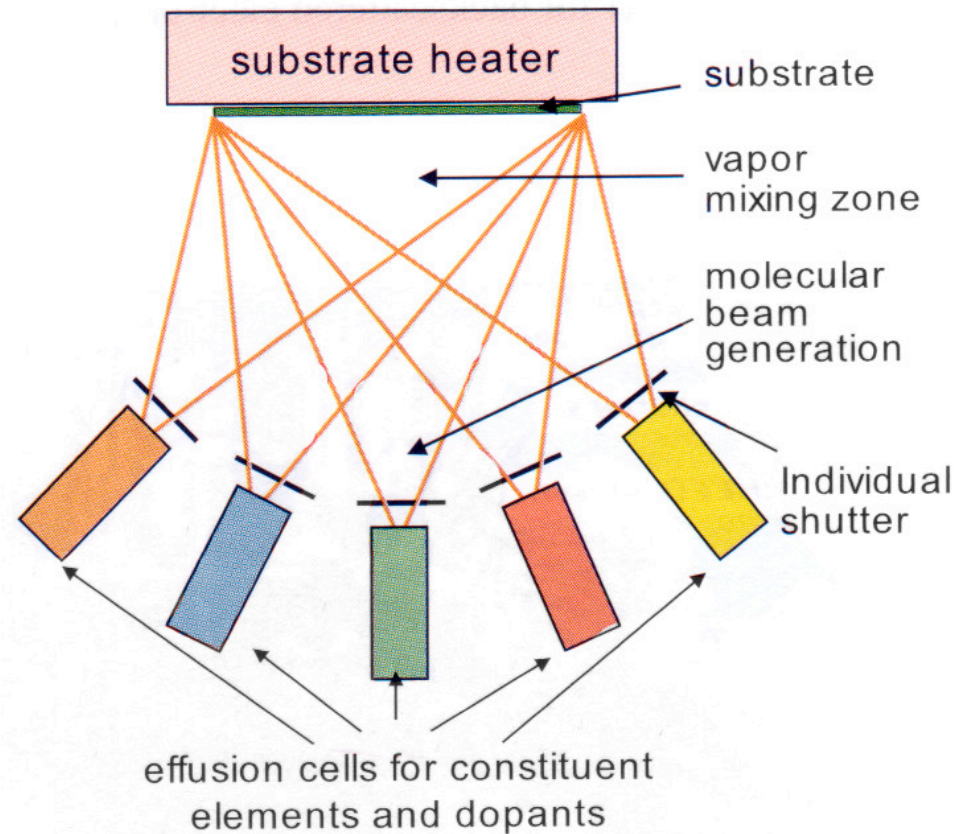
→ Layer by layer growth initially
→ When $d > d_{\text{crit}}$, island growth

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Molecular Beam Epitaxy (MBE)

- Developed by A. Cho (UIUC alum) in 1971.
- Constituent elements in the form of molecular beams are deposited onto a heated substrate
- Molecular beams are formed from thermally evaporated sources
- Atomic level control is achieved via shuttering
- Resulting material has abrupt interfaces and exceptional control over thickness, doping, and composition

MBE Schematic



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Characterization

(1) Electron diffraction → characterizes surface structure *in situ* during growth

(A) Low energy electron diffraction (LEED)

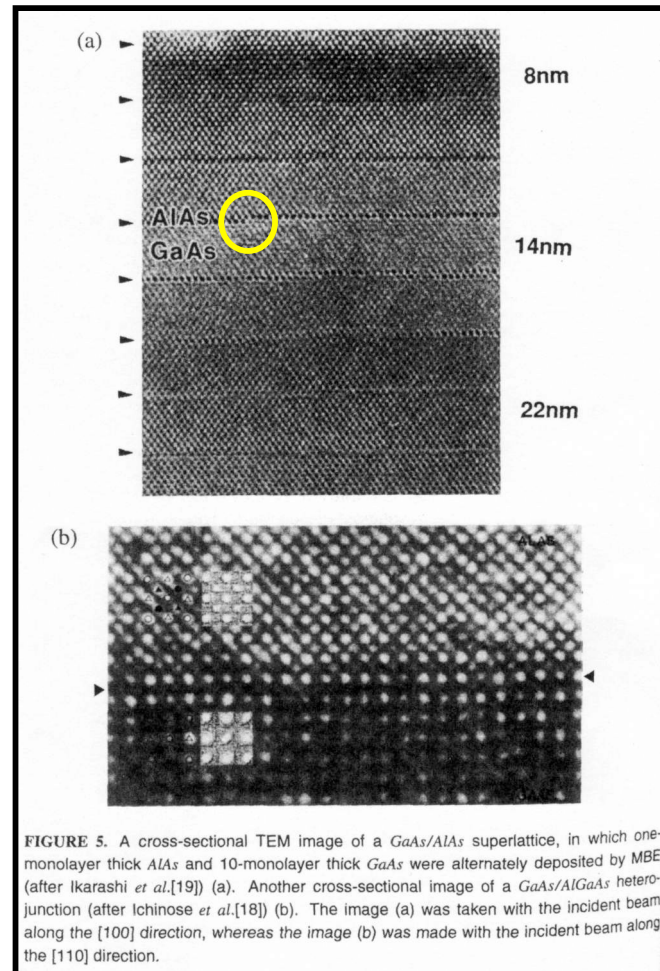
(B) Reflection high energy electron diffraction (RHEED)

(2) Cross-sectional microscopy → characterizes interfaces after growth

(A) Transmission electron microscopy (TEM) → structure and chemical composition

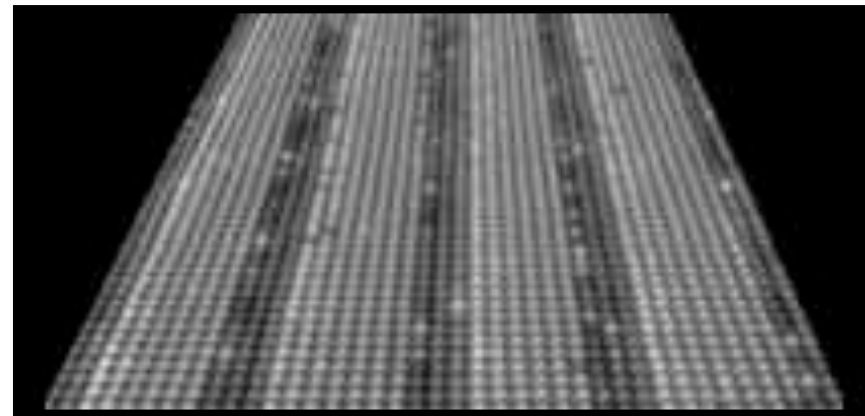
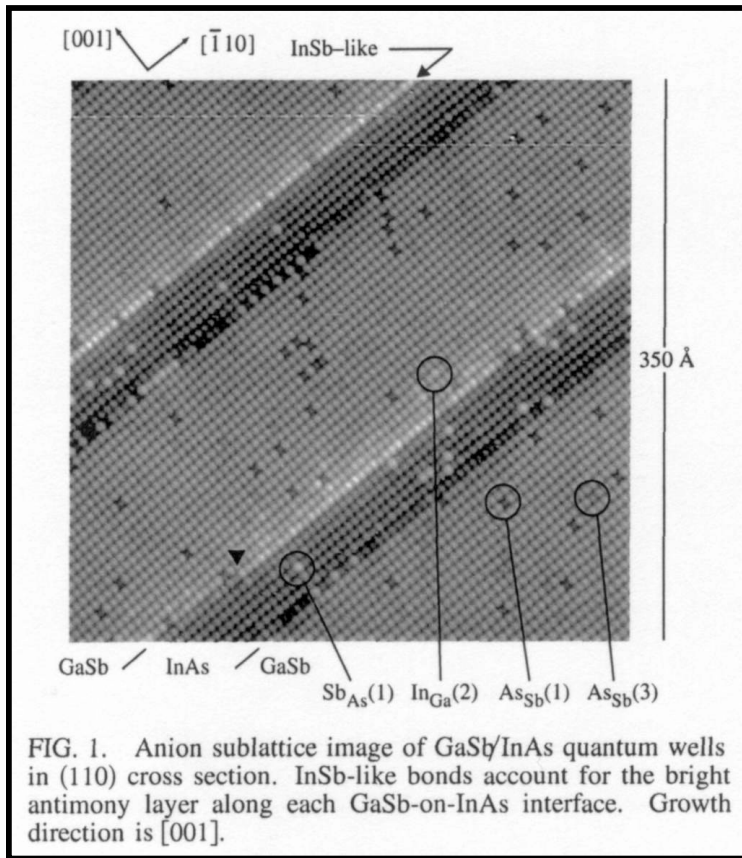
(B) Scanning tunneling microscopy (STM) → structure and electrical properties

Cross-sectional
TEM image of
GaAs/AlAs
superlattice:



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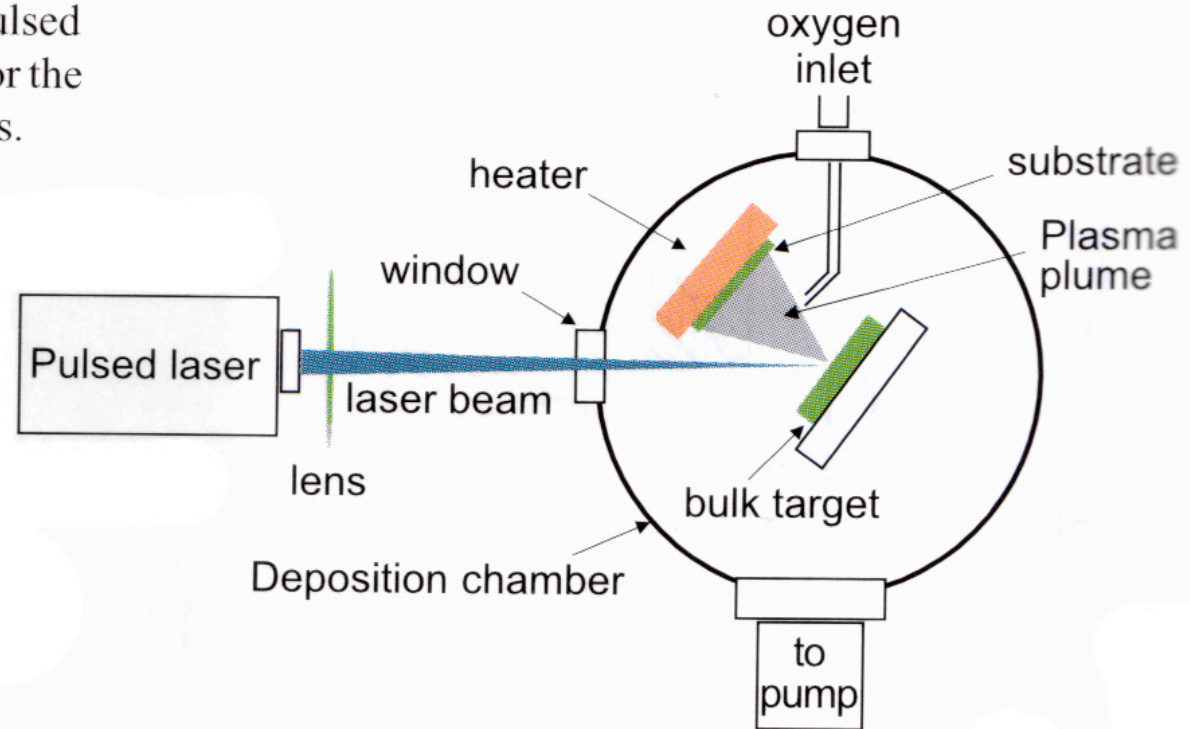
Cross-sectional STM images of GaSb/InAs superlattice:



J. Steinshnider, *et al.*, *Phys. Rev. Lett.*, **85**, 2953 (2000).

Pulsed Laser Deposition (PLD)

Figure 20: Set-up of a pulsed laser deposition system for the deposition of oxide layers.



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Time Evolution of the PLD Plasma Plume

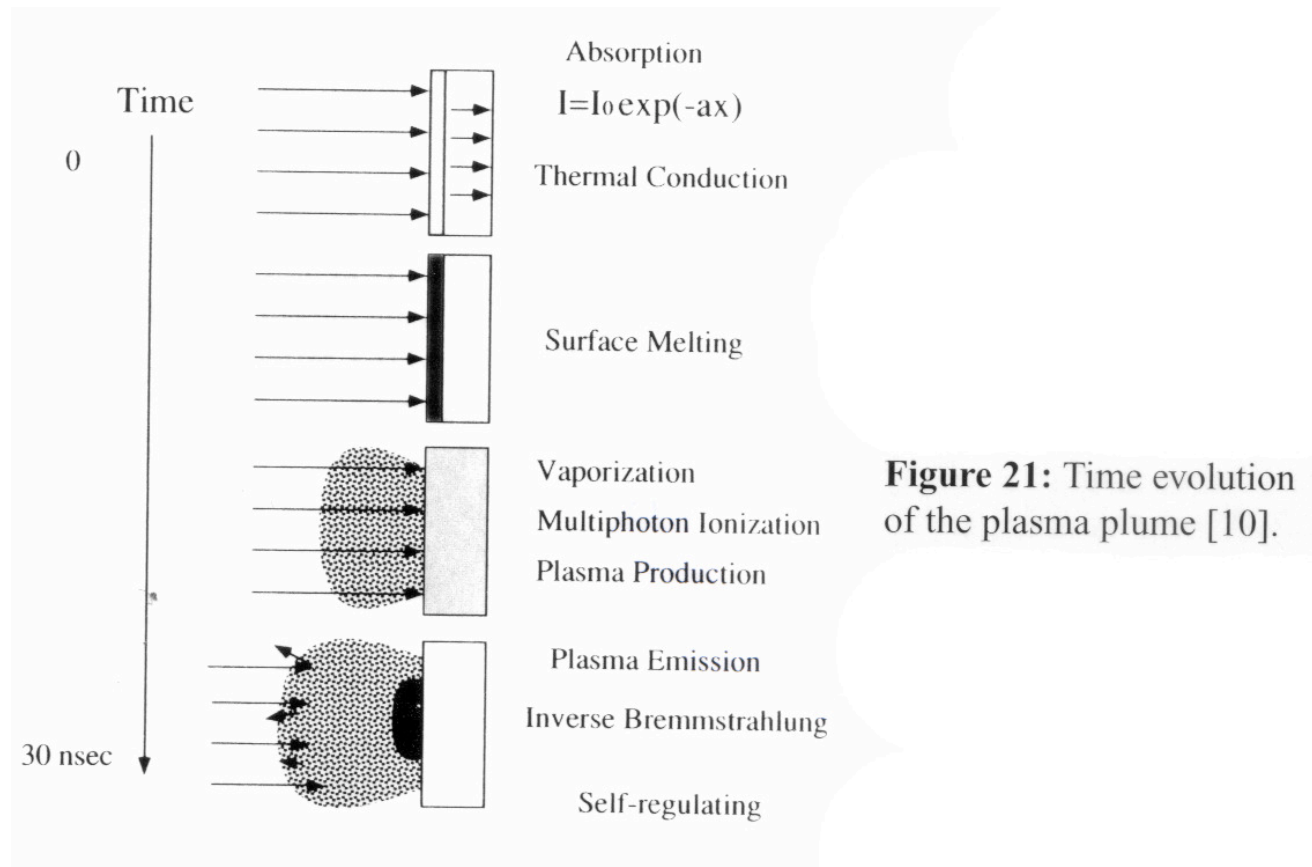
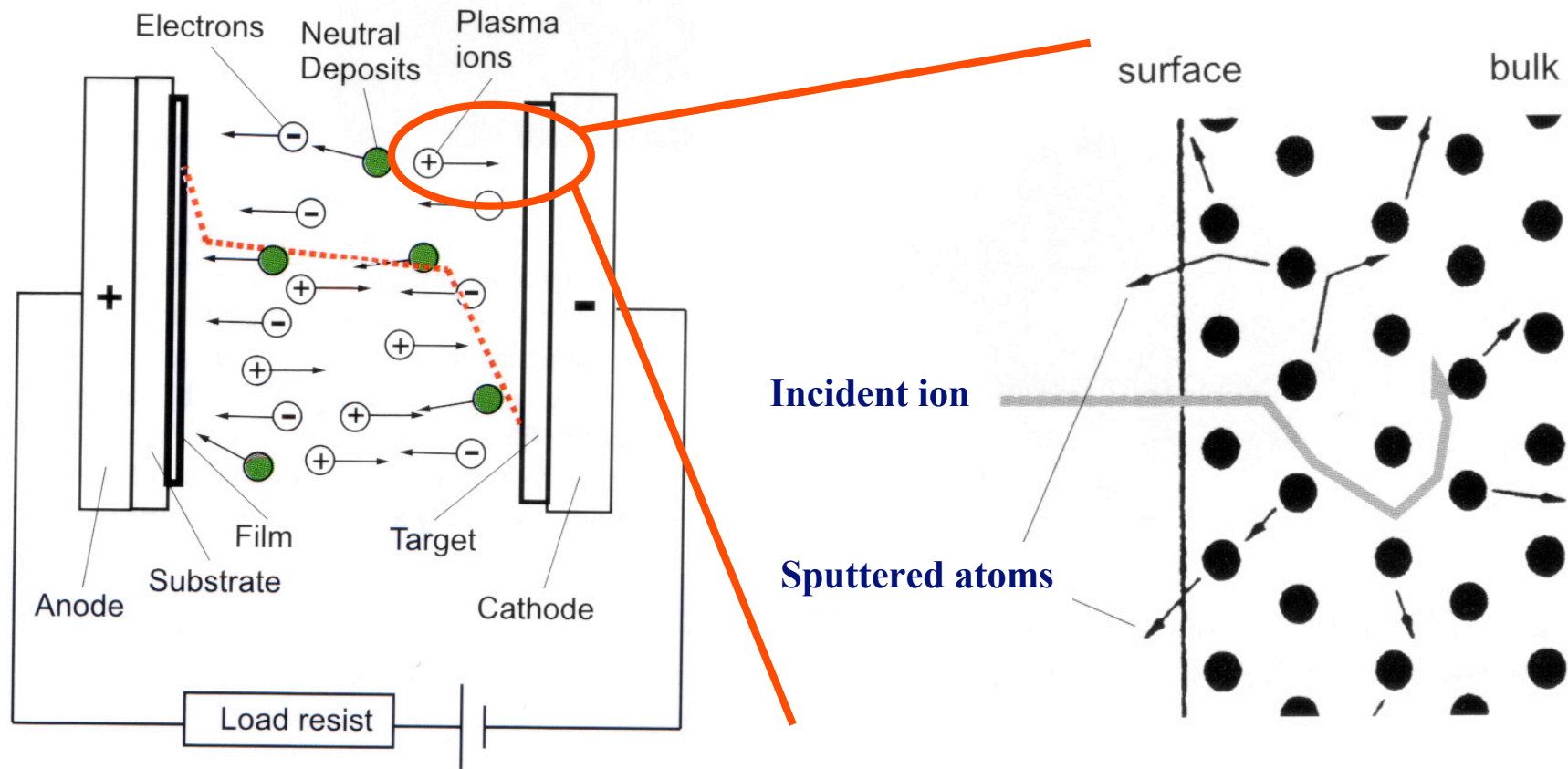


Figure 21: Time evolution of the plasma plume [10].

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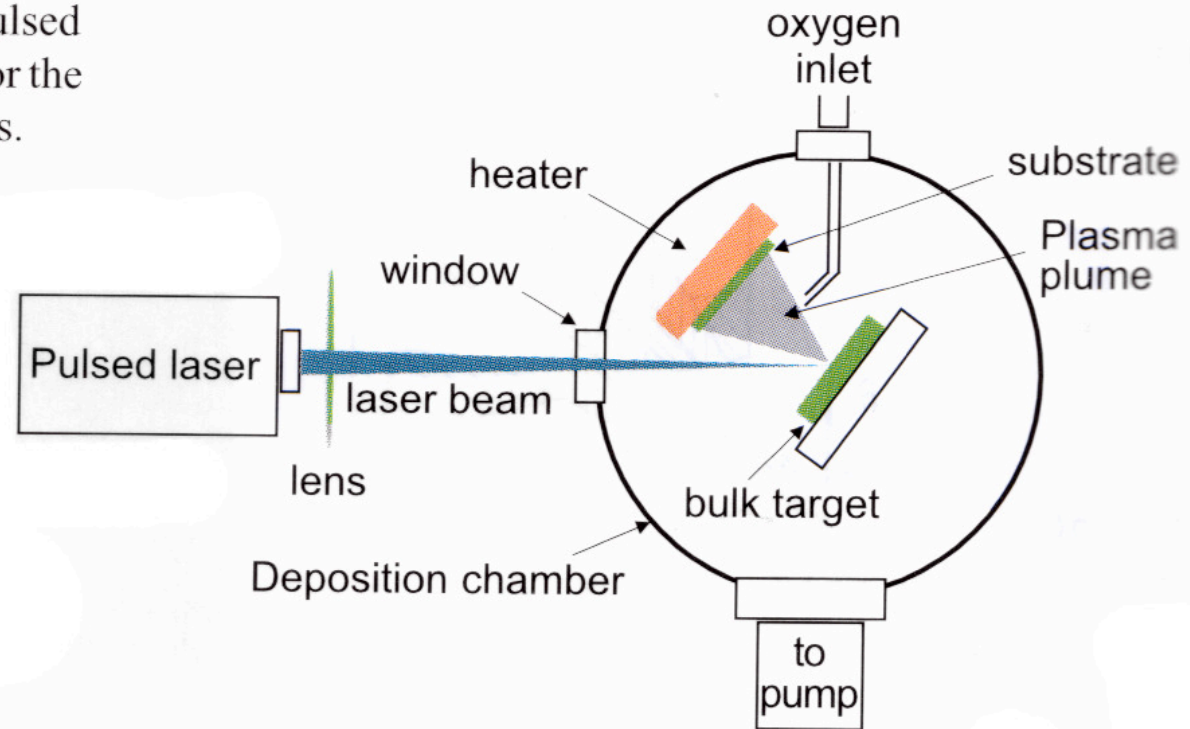
Sputter Deposition



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Pulsed Laser Deposition (PLD)

Figure 20: Set-up of a pulsed laser deposition system for the deposition of oxide layers.



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DC Sputter Deposition

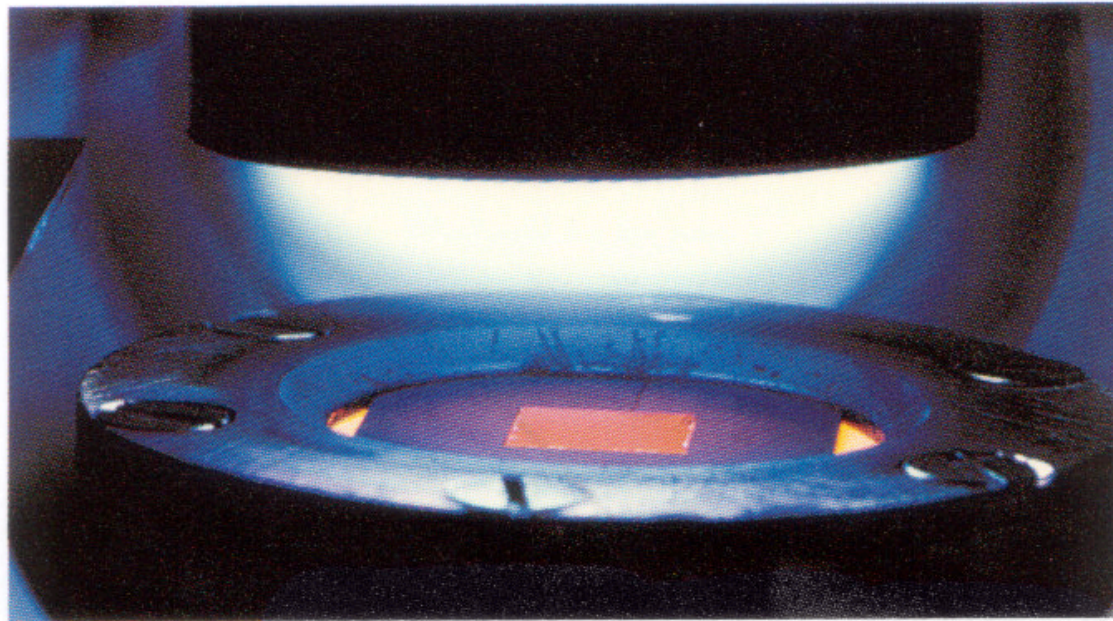
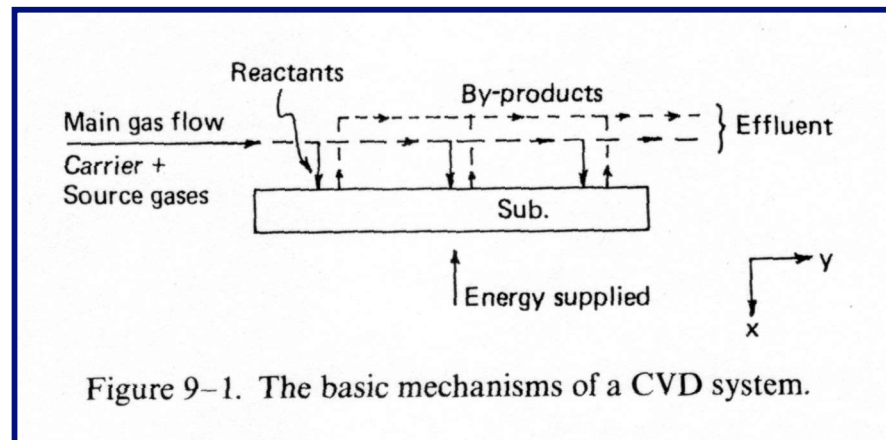


Figure 28: DC-sputter deposition in 'high pressure' oxygen.

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Chemical Vapor Deposition (CVD)

- Source materials are delivered to the surface in the gas phase
 - Energy is applied (heat, plasma, electromagnetic radiation)
- Chemical bonds break → Free radicals react with substrate



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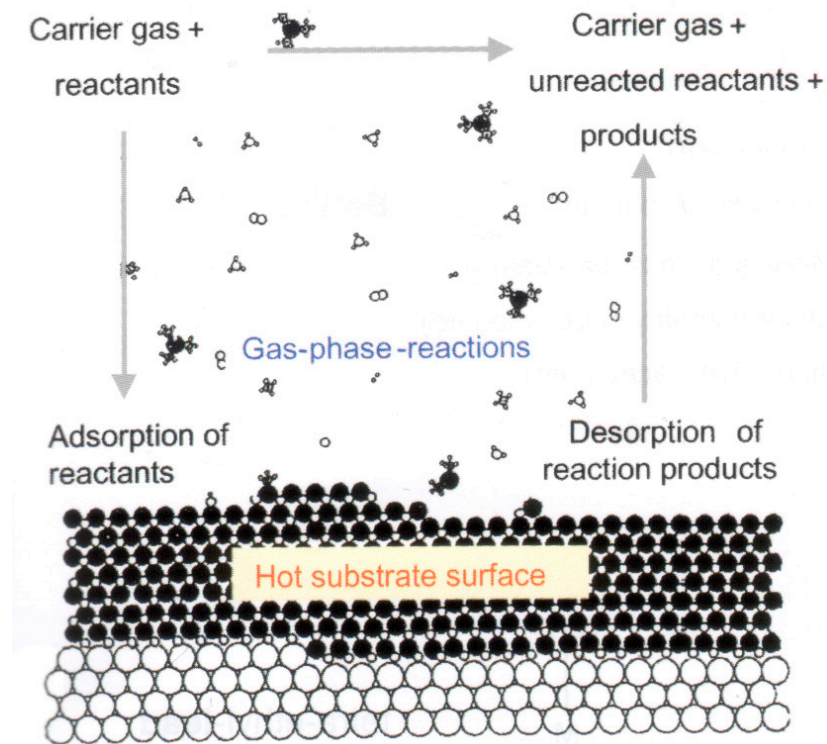
Table 9-1. Typical CVD Processes and Reactions

Pyrolysis:		
	$\text{SiH}_4\uparrow \rightarrow \text{Si}\downarrow + 2 \text{H}_2\uparrow$	(a)
	$\text{SiH}_2\text{Cl}_2\uparrow \rightarrow \text{Si}\downarrow + 2 \text{HCl}\uparrow$	(a')
Photolysis:		
	$\text{SiH}_4\uparrow + 2 \text{N}_2\text{O} \rightarrow \text{SiO}_2\downarrow + 2 \text{H}_2\uparrow + 2 \text{N}_2\uparrow$	(b)
Reduction:		
	$\text{SiCl}_4\uparrow + \text{H}_2\uparrow \rightarrow \text{Si}\downarrow + 4 \text{HCl}\uparrow$	(c)
	$\text{SiHCl}_3\uparrow + \text{H}_2\uparrow \rightarrow \text{Si}\downarrow + 3 \text{HCl}\uparrow$	(c')
Oxidation:		
	$\text{SiH}_4\uparrow + \text{O}_2\uparrow \rightarrow \text{SiO}_2\downarrow + 2 \text{H}_2\uparrow$	(d)
Reduction-Oxidation:		
	$3 \text{SiH}_4\uparrow + 4 \text{NH}_3\uparrow \rightarrow \text{Si}_3\text{N}_4\downarrow + 12 \text{H}_2\uparrow$	(e)
Doping Reactions:		
	$n \text{SiH}_4\uparrow + \text{B}_2\text{H}_6\uparrow \rightarrow n \text{Si}\downarrow + 2 \text{B}_2\downarrow + (2n + 3)\text{H}_2\uparrow$	(f)
	$n \text{SiH}_4\uparrow + 2 \text{PH}_3\uparrow + n \text{O}_2\uparrow \rightarrow n \text{SiO}_2 + 2 \text{P}\downarrow + (2n + 3)\text{H}_2\uparrow$	(g)

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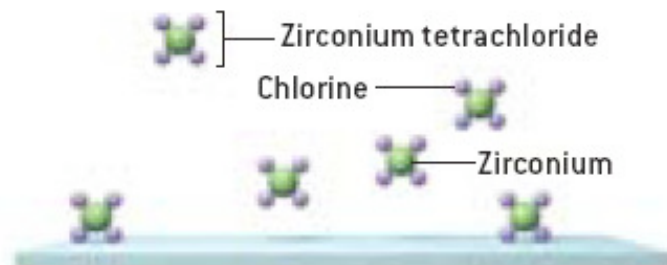
Metal-Organic CVD (MOCVD)

Relies on organometallic precursor compounds:



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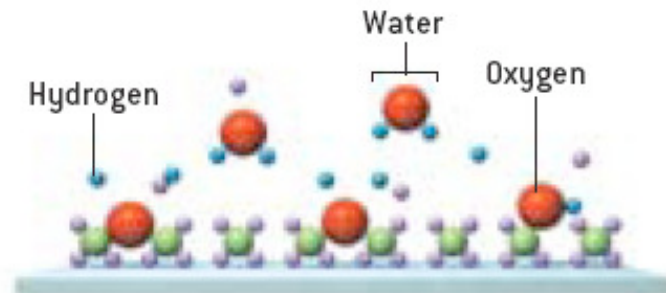
Atomic Layer Deposition



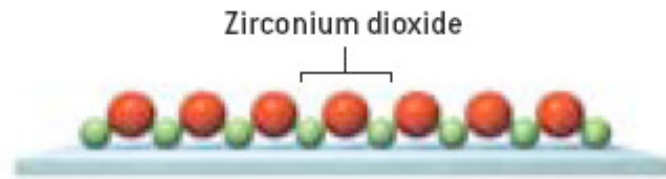
1 The surface is exposed to the first of two gases, here zirconium tetrachloride [ZrCl₄].



2 Molecules of ZrCl₄ adhere to the surface but not to one another.



3 The coated surface is exposed to a second gas, in this case steam (H₂O).



4 The ZrCl₄ on the surface reacts with the water (H₂O) to form a single-molecule-thick veneer of the desired material, zirconium dioxide [ZrO₂].

G. D. Hutcheson, *et al.*, *Scientific American*, **290**, 76 (2004).

Other Epitaxy Considerations

- In addition, defects in the underlying substrate will propagate into the growing film

(NOTE: Overgrowth can minimize defects – e.g., GaN on SiC)

- Periodic nanoscale defects will induce nanostructures in subsequent film growth. For example,

- (1) Atomic steps
- (2) Pre-patterned substrates

→ Although epitaxy always gives nanoscale control over 1-D, it can also provide 2-D and even 3-D control.

Epitaxial Lateral Overgrowth (ELOG)

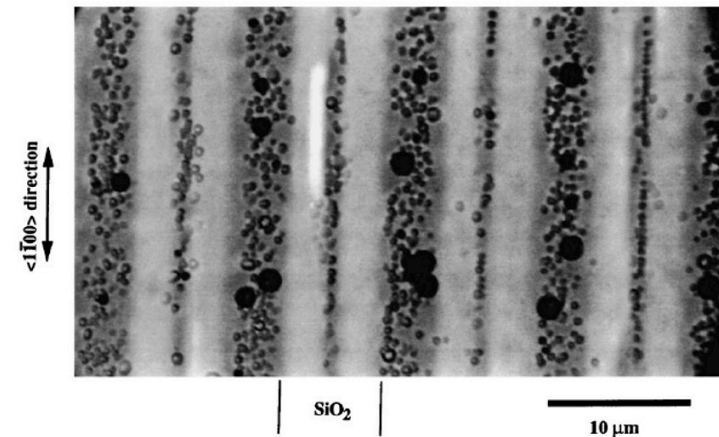
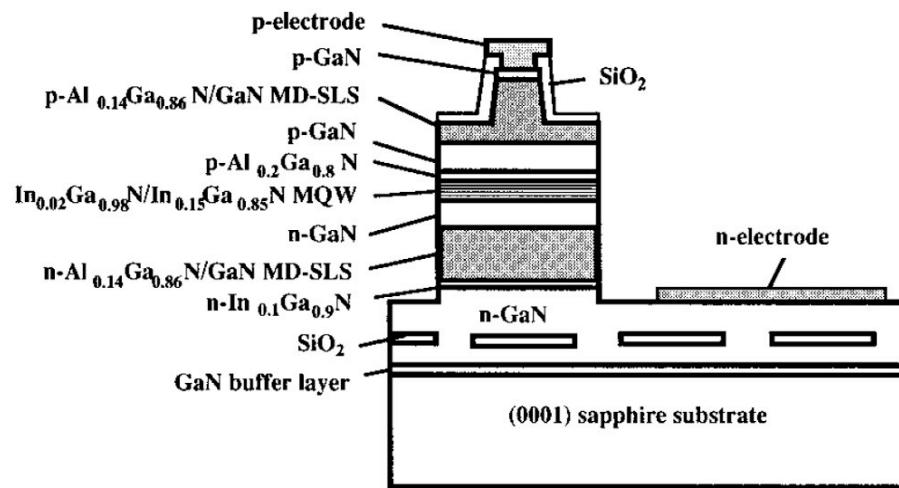


FIG. 1. Surface morphology after 4- μm etching of the ELOG substrate. The SiO₂ stripe width and window width were 7 μm and 4 μm , respectively. The etch pit density was about $2 \times 10^8 \text{ cm}^{-2}$ in the region of the 4- μm -wide stripe window.

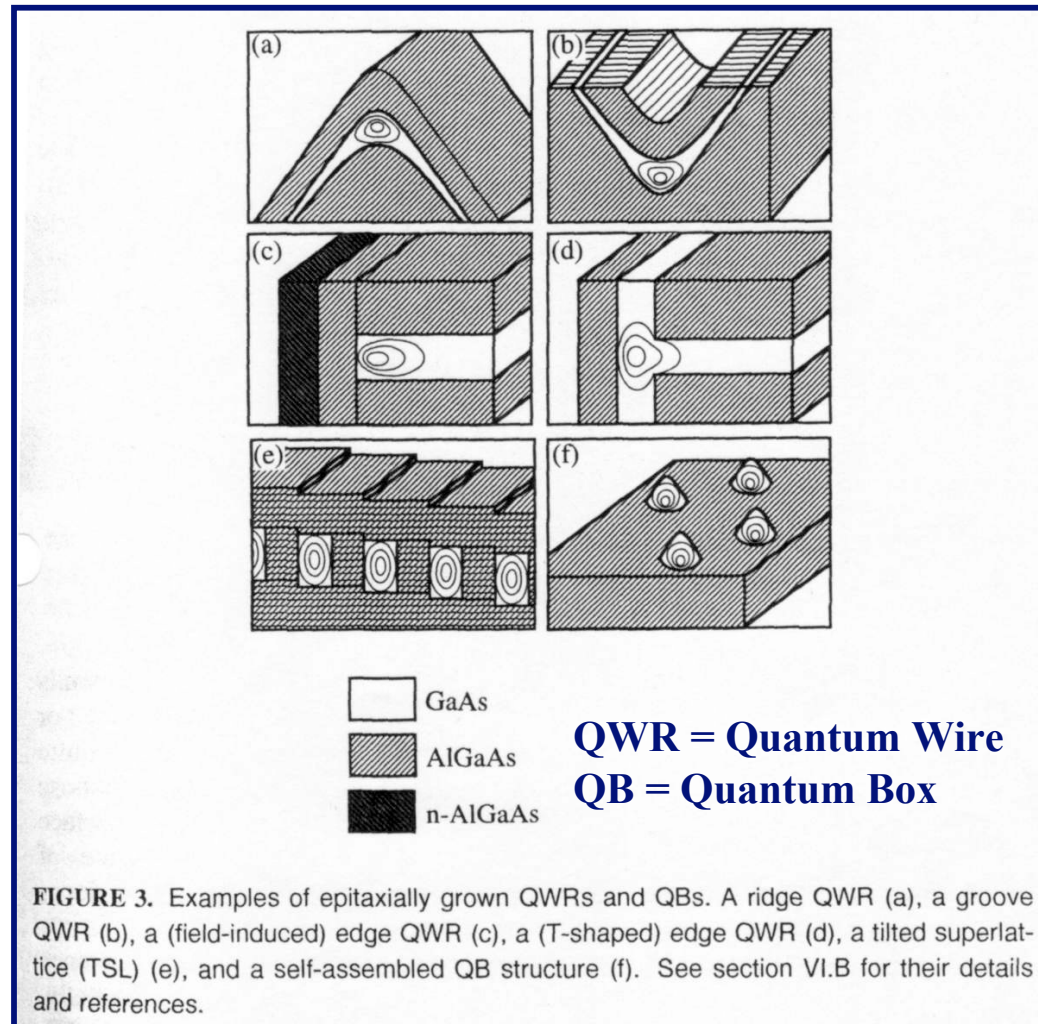
Threading dislocations are dramatically reduced in the overgrowth region.

S. Nakamura, *et al.*, *Appl. Phys. Lett.*, **72**, 211 (1998).

2-D and 3-D Control by Epitaxy

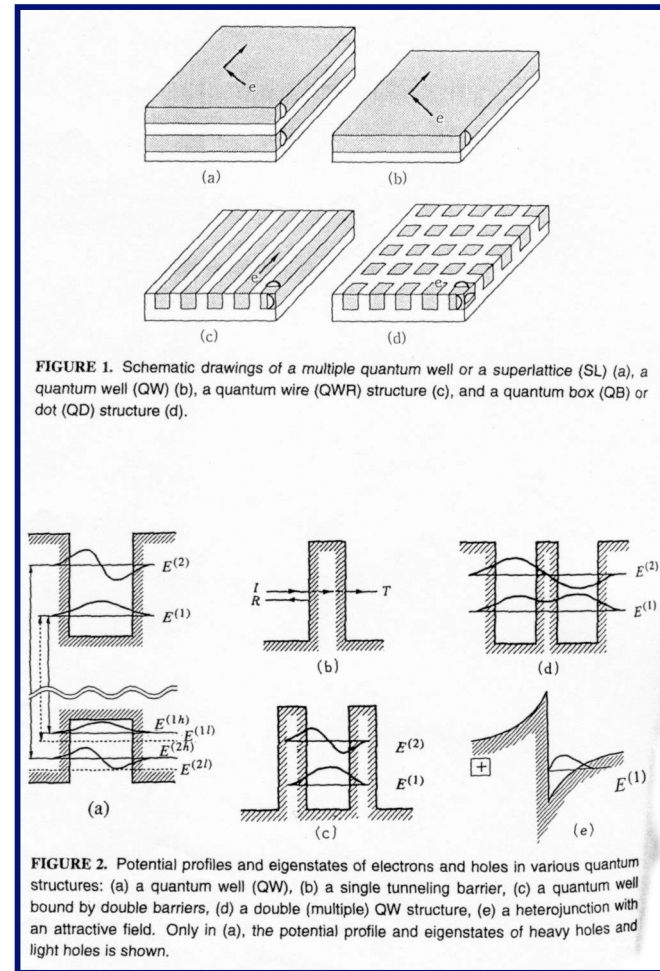
Examples of structures grown by epitaxy:

- (1) Ridge quantum wires → electrons confined to thicker regions (lower energy)
- (2) Edge quantum wires → growth on cleaved heterostructure
- (3) Self-assembled dots → form due to lattice mismatch in 2-D
- (4) Self-assembled nanowires → form due to lattice mismatch in 1-D



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NOTE: Atomically controlled heterostructures have unique electronic and optical properties due to control of bandstructure:



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Quantum Mechanical Effects

1-D Time-independent Schrödinger Equation:

$$-\frac{\hbar^2}{2m} \frac{d^2\psi}{dz^2} + V\psi = E\psi$$

(1) Quantization of energy levels in wells

(2) Tunneling through barriers

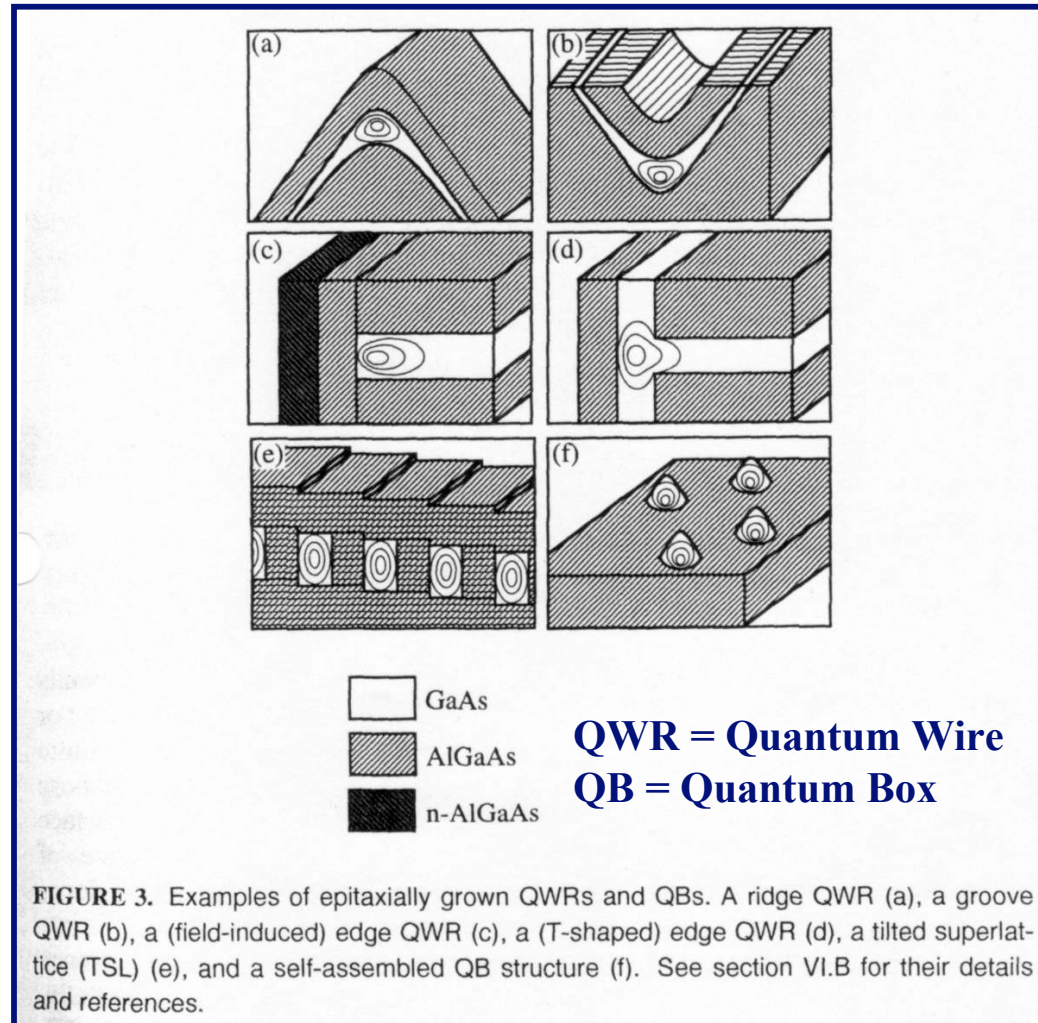
→ When heterostructures are monolayers thick, atomic level defects have a serious impact

Quantum Mechanical Property Changes

Wells and tunnel barriers allow for quantum effects:

- (1) Tailored optical properties
- (2) Resonant tunneling diodes
- (3) Single electron devices
- (4) 2-dimensional electron gas

→ We will revisit these applications later in the course



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Patterned substrates:

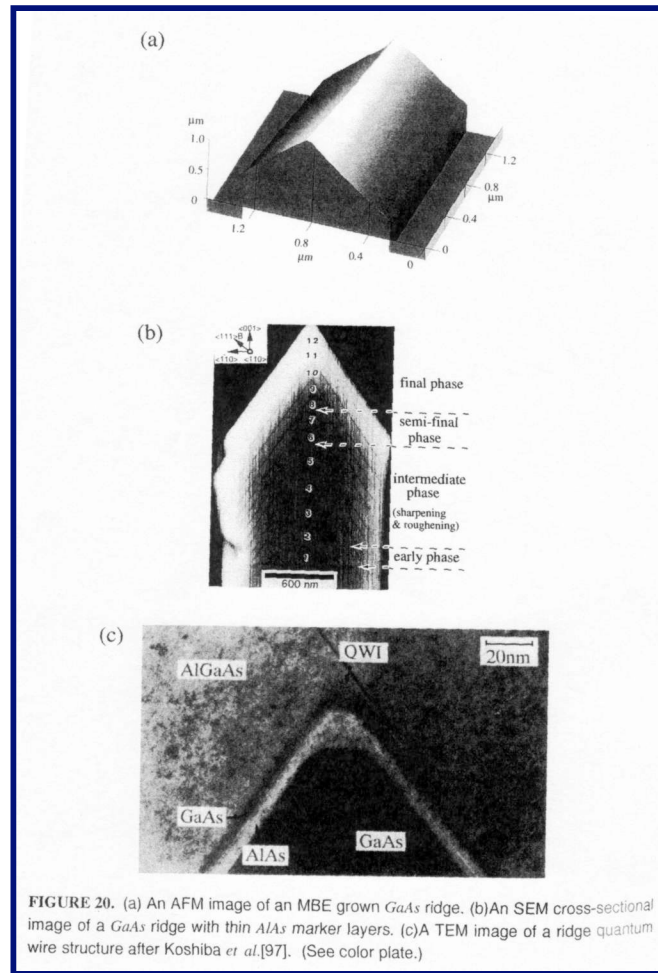
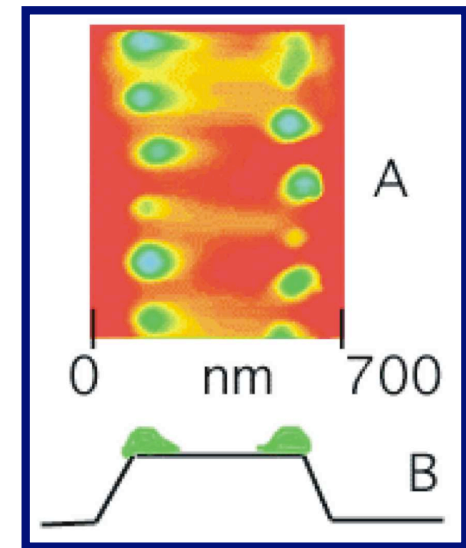


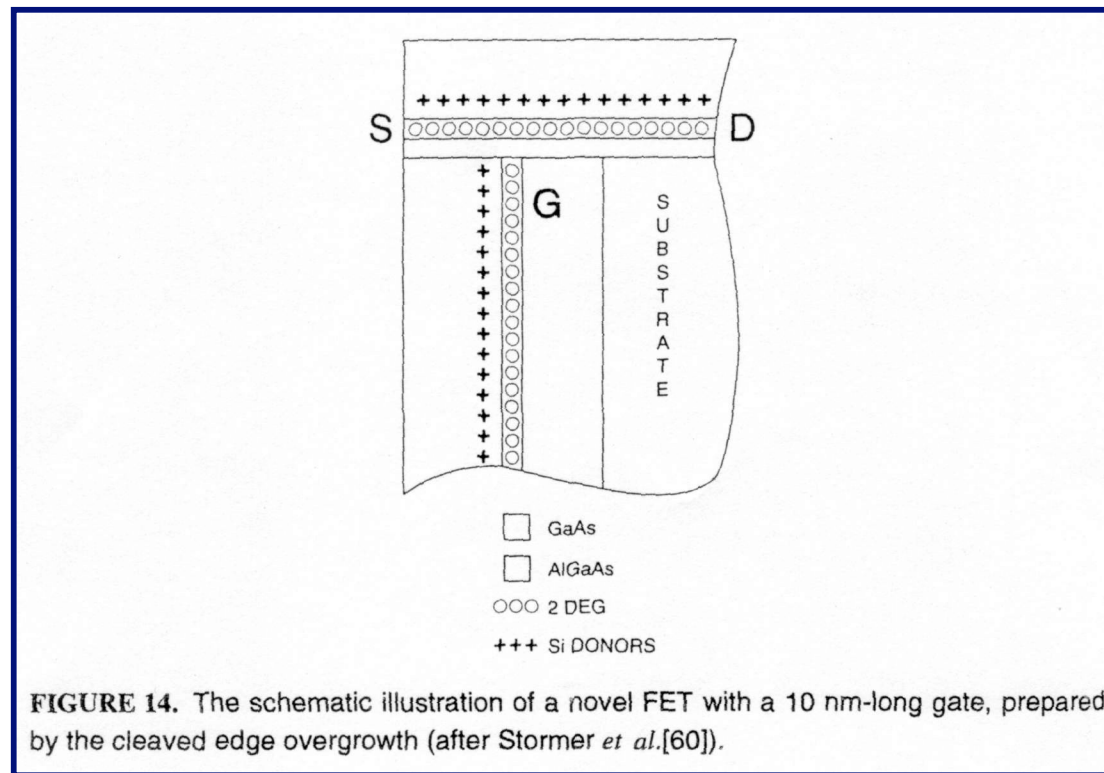
FIGURE 20. (a) An AFM image of an MBE grown *GaAs* ridge. (b) An SEM cross-sectional image of a *GaAs* ridge with thin *AlAs* marker layers. (c) A TEM image of a ridge quantum wire structure after Koshiba *et al.*[97]. (See color plate.)



Courtesy of M. Bedzyk

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Cleaved Edge Overgrowth



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Self-assembled quantum dots (Ge on Si(001))

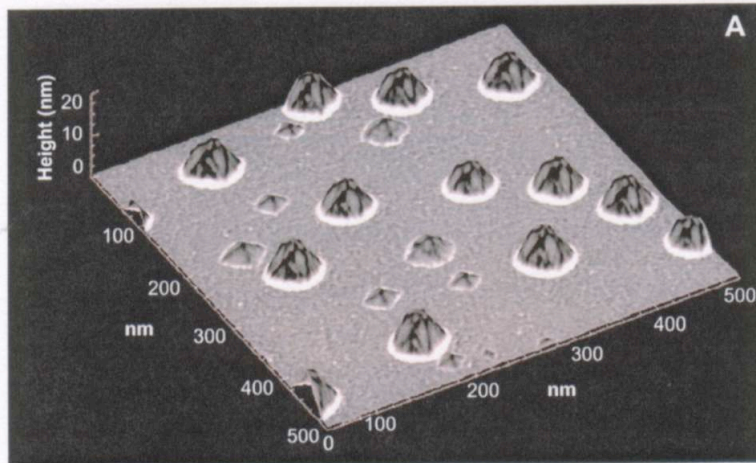
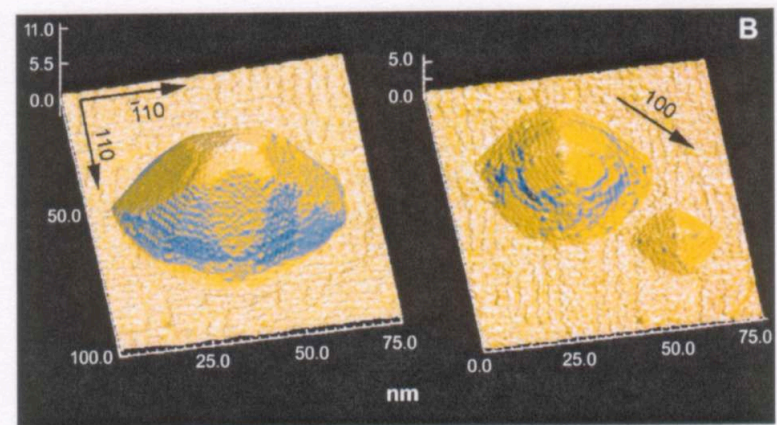


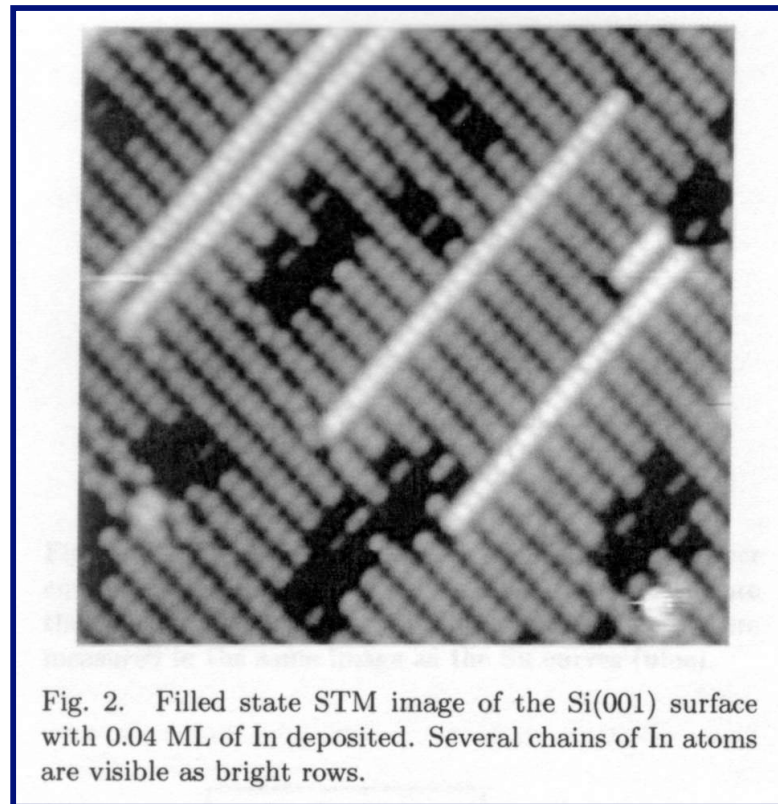
Fig. 1. (A) STM topograph [surface height $h(x, y)$ as a function of position] of strained Ge nanocrystals on Si(001), showing both pyramids and domes. The gray scale is proportional to the local surface curvature as determined by the Laplacian $\nabla^2 h(x, y)$: positive curvature is white, flat areas are gray, and negative curvature is black. This visualization mode emphasizes the nano-



crystal edges, which appear black and clearly frame the gray facets of the nanocrystals, showing that edges contain a significant fraction of the atoms on the surface of the nanocrystals. **(B)** Higher magnification images of the nanocrystals: (left) a mature dome and (right) a nanocrystal entering the transition stage and a small pyramid.

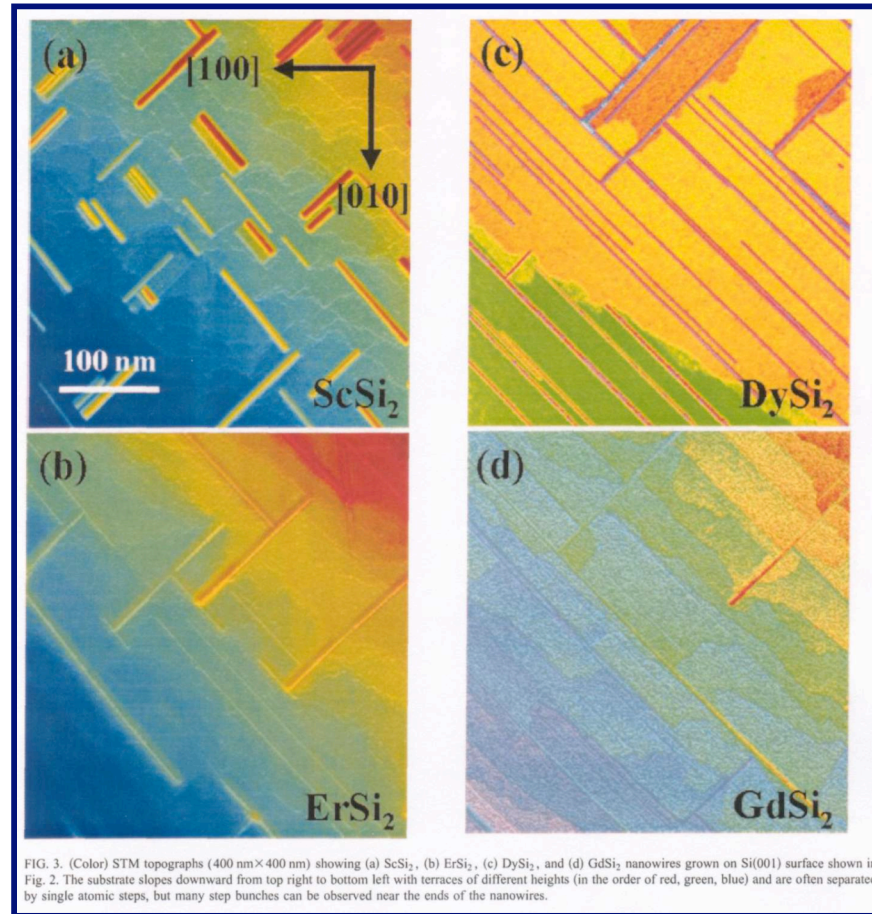
G. Medeiros-Ribeiro, *et al.*, *Science*, **279**, 353 (1998).

Self-assembled atomic wires (In on Si(001))



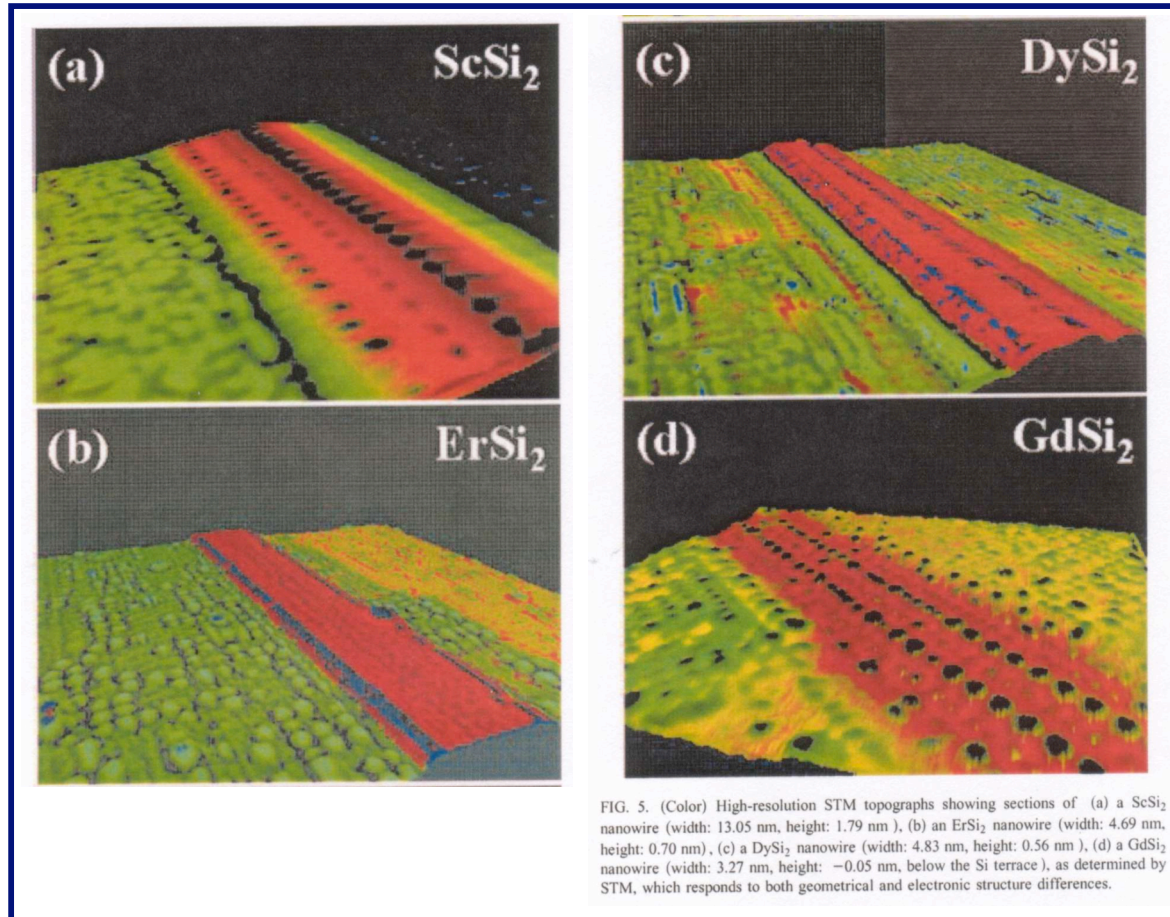
J. Nogami, *et al.*, *Surface Review and Letters*, **7**, 555 (2000).

Self-assembled nanowires (silicides on Si(001))



Y. Chen, *et al.*, *J. Appl. Phys.*, **91**, 3213 (2002).

Self-assembled nanowires (silicides on Si(001))



Y. Chen, *et al.*, *J. Appl. Phys.*, **91**, 3213 (2002).

Self-assembled nanowires (silicides on Si(001))

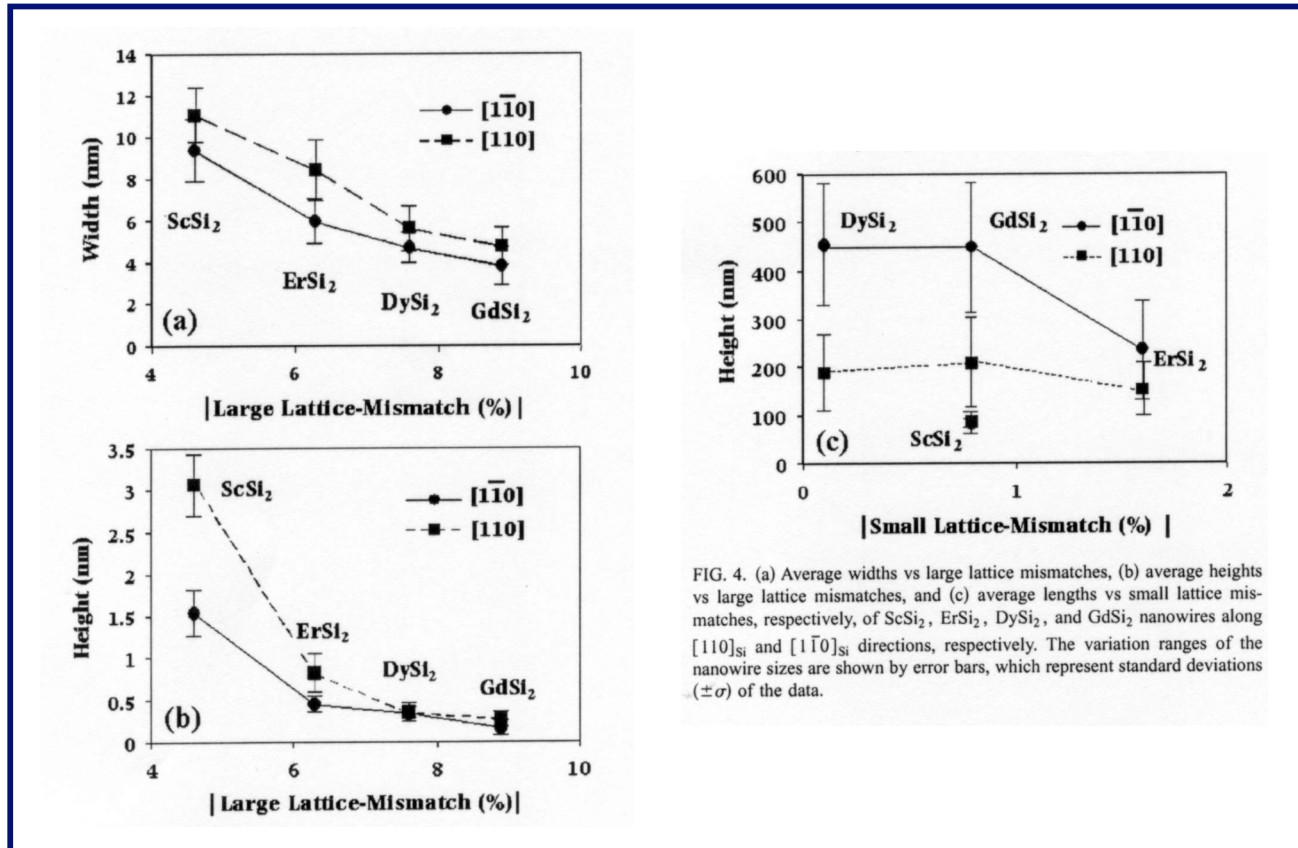


FIG. 4. (a) Average widths vs large lattice mismatches, (b) average heights vs large lattice mismatches, and (c) average lengths vs small lattice mismatches, respectively, of ScSi₂, ErSi₂, DySi₂, and GdSi₂ nanowires along [110]_{Si} and [1 $\bar{1}$ 0]_{Si} directions, respectively. The variation ranges of the nanowire sizes are shown by error bars, which represent standard deviations ($\pm\sigma$) of the data.

Y. Chen, *et al.*, *J. Appl. Phys.*, **91**, 3213 (2002).